

Title: Li^+-Li^+ and Na^+-Li^+ ion pairs in aqueous solution

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Abstract: Like ion pair Li^+-Li^+ and mix ion pair Na^+-Li^+ in aqueous solution were investigated by high level ab initio calculation and quantum mechanical effective fragment potential molecular dynamics (QM/EFP MD). Our QM/EFP-MD predicts that like ion pair Li^+-Li^+ are more likely to be observed than mix ion pair Na^+-Li^+ in the solution. In a particular, the formation of Li^+-Li^+ is less than that of Na^+-Na^+ , and the formation of Na^+-Li^+ is even worse. The local structural analysis indicates that three-water bridging structure is dominating in Li^+-Li^+ ion pair in the solution whereas the Na^+-Li^+ ion pair mostly bound to one water molecule. Both the MD and QM study suggests that one water bridging structure is very unstable.

Keywords: Ion pairing, Li^+-Li^+ ion pair, Na^+-Li^+ mixed ion pair, Ab initio calculations

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